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The Use of Thermal Sprayed Zinc Alternative to Hot Dipped Galvanizing

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INTRODUCTION

Galvanizing has a historical background of use in reducing corrosion on piping products.¹ Hot dipped galvanizing results in a cathodic protection system utilizing zinc as a sacrificial anode material dispersed on the surface of iron or steel products.

Hot dipped galvanizing has been used in various markets for over 100 years as a method of corrosion protection. The basic steps to the process are as follows:

1. Parts are degreased in a hot alkaline solution.
2. Surface rust and scale are removed.
3. Parts are immersed in a liquid flux to prevent oxidation prior to dipping in the molten zinc.
4. Parts are immersed in the molten bath of zinc at a temperature around 850 °F.
5. The parts are withdrawn from the galvanizing bath and the excess zinc is removed by draining or centrifuging and allowed to cool and dry.
6. Parts are inspected to complete the process.

In the hot dipped galvanizing method, it is important to keep the temperature of the bath above the temperature of molten zinc. This requires a large fuel consumption and a considerable loss of zinc occurs. Increased inspection and rework due to zinc drips may be required on complicated shapes such as grooved couplings and fittings.

An alternative to the hot dipped galvanizing process is the use of Thermal Sprayed Zinc. Like galvanizing, Thermal Sprayed Zinc also provides a cathodic protection system and is replacing hot dipped galvanizing in many applications for reasons of effectiveness and economics.

Thermal spray coatings are widely used in preventing corrosion of many materials. A common application is the use of zinc to protect iron substrates. In this method, the surface of the metal is prepared by abrasive blast cleaning to ensure it is cleaned and roughened. Metal wire or powder is fed at a controlled rate into a flame or arc spraying process and the atomized metal impinges upon the surface being coated and becomes bonded to it. This process is easily adapted to production environments and is especially suited to complicated surfaces found in grooved couplings and fittings.

¹ It is the Designer's responsibility to select products suitable for the intended service and to ensure that materials are acceptable for the specific application.

CATHODIC PROTECTION

All metals have properties that cause them to react as an anode or a cathode when coupled to another material in a corrosive environment. The determination as to how two metals will react when coupled in a corrosive environment is based on a galvanic series shown below.

	<u>Galvanic Series</u>
Anodic	Magnesium
	Zinc
	Cadmium
	Aluminum
	Iron
	Steel
	Stainless Steels (active)
	Brass
	Copper
	Monel
	Lead
	Tin
	Silver
	Gold
Cathodic	Platinum

The application of zinc (anodic coating) to iron substrates forms a protective layer and results in a corrosion protection referred to as cathodic protection or sacrificial protection. The substrate iron becomes the cathode and the zinc coating becomes the sacrificial anode.

ABOUT THE AUTHOR

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Mr. Radzik's interests include the research, design and applications of Grooved and CPVC piping products. He has been involved in piping products for more than 25 years. Mr. Radzik's engineering and applications knowledge of piping products gave him the opportunity to work as Senior Project Engineer and later as Engineering Manager for Central Sprinkler Company, where he was responsible for the technical development of their grooved product line. Mr. Radzik is currently the Director of Engineering for Tyco Fire & Building Products, which includes the Central Grooved Piping Products and Tyco CPVC Piping Products brands. His responsibilities include the engineering and quality assurance of Grooved and CPVC piping products worldwide.

In addition to his research into piping products, Mr. Radzik has authored published articles in "Heating/Piping/Air Conditioning", "FPC/Fire Protection Contractor", "American Society Of Sanitary Engineering News Bulletin" and "Plant Engineering". He continues his involvement in the industry with memberships including The American Society of Mechanical Engineers, National Fire Protection Association, National Association of Corrosion Engineers and American Water Works Association.

Mr. Radzik is also named on various patents both issued and pending.

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